26. Synthesis of Thiazole and Fused Thiazolo Derivatives¹)

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The syntheses of thiazole and fused thiazolo derivatives 2-4, 6-8, 10a-11b, 13-16 from heterocyclic isothiocyanates 1, 5, 9, and 12 bearing an ortho ester group and bifunctional reagents, such as substituted propargylamines, is described. Different regioselectivity of intramolecular nucleophilic attack of the thiourea S-atom on the $C \equiv C$ bond, resulting in the formation of both thiazolo and thiazino derivatives, as well as NMR structure elucidation are discussed.

Introduction. – A variety of methods are available for the synthesis of thiazole derivatives [1], but there are only few syntheses leading to fused tricyclic thiazolo heterocycles [2]. Some fused thiazolo heterocycles possess biological or pharmacological activities [3] [4].

The syntheses of thiazole derivatives using a reaction between thiourea and a compound containing a primary halogen group adjacent to a $C \equiv C$ bond has been already described [5]. High regioselectivity for this ring closure has been described with the formation of only a five-membered ring.

The cyclization reactions of thiourea derivatives prepared by the reaction of the secondary α, α -disubstituted propargylamines with isothiocyanates led to the formation of only tetrahydrothiazole derivatives [6]. This method has been also used for the syntheses of thiazolo-quinazole derivatives from methyl 2-(isothiocyanato)benzoate, and no formation of thiazino ring was observed [7]. Heterocyclic isothiocyanates bearing ester groups in *ortho*-positions are useful syntheses of the syntheses of fused *N*-heterocycles by addition of substituted amines and related compounds [3] [8–11].

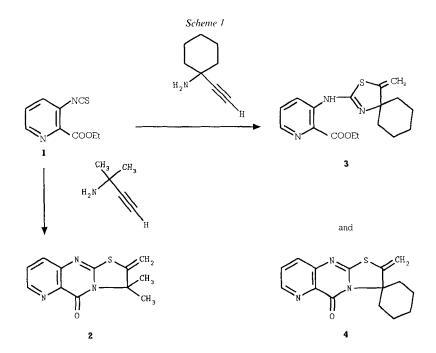
In the present paper, we describe the syntheses of isomeric thiazolo[3,2-a]pyridopyrimidine, thiazolo[2,3-b]- and thiazino[2,3-b]pteridine, and thiazolo[3,2-a]thieno-[3,2-d]pyrimidine derivatives.

Results and Discussion. – The reaction between heterocyclic isothiocyanates and substituted propargylamines proceeds in three steps. The first step is the formation of a thiourea derivative, which could not be isolated. The second step is an intramolecular nucleophilic attack of the thiourea S-atom onto the C \equiv C bond.

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Regioselectivity in a ring closure by internal nucleophilic attack has been described [5] [12]. For nucleophilic attack of the S-atom at the C \equiv C bond, the bonding distances are close enough for both five- and six-membered ring formation [13], and according to *Baldwin*'s ring closure rules, both are favored [12]. We observed different regioselectivity of nucleophilic attack of the S-atom at the C \equiv C bond, and the formation of both thiazolo[2,3-*b*]- and thiazino[2,3-*b*]pteridine derivatives (*cf. Scheme 4*). In other reactions, thiazino derivatives were formed in a small amount, according to ¹H-NMR spectra. In some cases, we could not separate both products. The third step is further condensation of thiazolo and/or thiazino derivatives to fused systems.

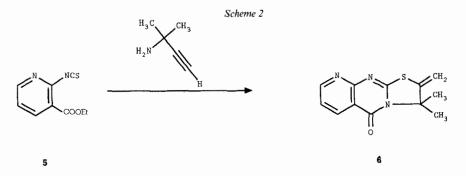
Ethyl 3-isothiocyanatopyridine-2-carboxylate (1) [9] reacted with 1,1-dimethyl-2propynylamine to 2,3-dihydro-3,3-dimethyl-2-methylidene-5-oxopyrido[3,2-d]thiazolo-[3,2-a]pyrimidine (2) and with eth-1-ynylcyclohexylamine to compounds 3 and 4 (*Scheme* 1). Both reactions formed five-membered rings, which could be isolated and detected



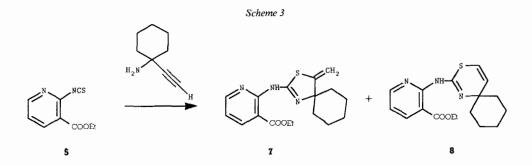
with NMR spectroscopy. Two *dublets* with a ${}^{2}J$ values of *ca*. 2.5 Hz are characteristic for an exocyclic methylidene group. In the NMR spectrum of the evaporated reaction mixture, we observed also signals for the thiazino isomer which was not isolated (*ca*. 5%).

An analogous treatment of ethyl 2-isothiocyanato-3-pyridinecarboxylate (5) [9] with 1,1-dimethylpropargylamine gave 2,3-dihydro-3,3-dimethyl-2-methylidene-5-oxopyrido-[2,3-d]thiazolo[3,2-a]pyrimidine (6; *Scheme 2*) and in a very little amount the thiazino derivative (*ca.* 5%) which was observed in the ¹H-NMR spectrum of the crude reaction product and was not isolated and purified. For the thiazino ring, two *dublets* of low

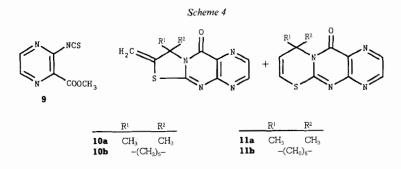
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intensity with a ${}^{3}J$ value of *ca*. 10 Hz at 5.65 and 6.09 ppm were observed. In the reaction of eth-1-ynylcyclohexylamine with isothicyanato-ester **5**, the expected fused product was not formed. In this case, we isolated only the thiazolo derivative **7** and the thiazino derivative **8** in ratio 1.78:1 (*Scheme 3*).

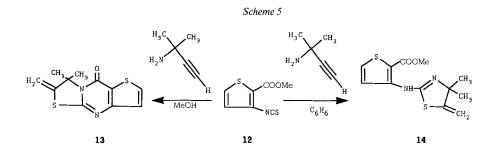


Methyl 3-isothiocyanatopyrazine-2-carboxylate (9) [10] reacted with substituted propargylamines with low regioselectivity in the second reaction step (*Scheme 4*) leading to a mixture of both five- and six-membered fused rings, *i.e.* thiazolo[2,3-*b*]pteridine derivatives **10a** and **10b** and thiazino[2,3-*b*]pteridine derivatives **11a** and **11b** which could not be separated chromatographically. The ratio of the products **10a/11a** is according to ¹H-NMR *ca*. 1:1.2, and for **10b/11b** the ratio is 3.6:1.

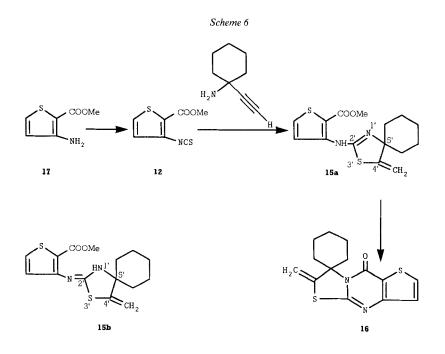


For the thiazino compounds, we observed two *dublets* with a ${}^{3}J$ value of *ca*. 10.6 Hz which is characteristic for olefinic protons (see *Exper. Part*).

When methyl 3-isothiocyanatothiophene-2-carboxylate (12) [3] was heated with 1,1dimethylpropargylamine in MeOH, it underwent a transformation to give 6,7-dihydro-7,7-dimethyl-6-methylidene-9-oxothiazolo[3,2-*a*]thieno[3,2-*d*]pyrimidine (13), which is a less known ring system [4]. In benzene, the reaction product was methyl 3-[(4,5-dihydro-4,4-dimethyl-5-methylidenethiazol-2-yl)amino]thiophene-2-carboxylate (14; Scheme 5).



Isothiocyanate 12 reacted with eth-1-ynylcyclohexylamine at elevated temperature, and prolonged reaction in MeOH or benzene led to methyl 3-[(4-methylidene-3-thia-1-azaspiro[4.5]dec-1-en-2-yl)amino]thiophene-2-carboxylate (15a). Further reaction of 15a with*t*-BuOK led to 6',7'-dihydro-6'-methylidenespiro[cyclohexane-1,7'-thiazolo[3,2-*a*]-thieno[3,2-*d*]pyrimidin]-9'-one (16; Scheme 6).



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The prototropic tautomerism in heterocyclic compounds involves the shift of a proton between a ring N-atom and a substituent atom connected to the ring. The relative position of a functional group with respect to the hetero atoms often determines the tautomeric nature of such compounds [14]. Dihydrothiazoles, bearing protomeric amino group in a position corresponding to enanime function, are in tautomeric equilibrium with this functional group [2]. In the case of the dihydrothiazoles **3**, **7**, **14**, and **15**, a proton shift between two N-atoms forming an amidine-like substructure with an endocyclic or exocyclic C=N bond has to be discussed.

Previously, we used the selective heteronuclear ${}^{13}C{}^{1}H$ -NOE technique for the analysis of the tautomerism in the pteridine system [10], and now we have applied this NMR technique for the determination of the tautomerism of compound 15 in a solution.

The assignment of the ¹³C-NMR spectrum of **15** was achieved by comparison of the ¹³C-chemical shifts of methyl 3-aminothiophene-2-carboxylate (**17**) [15]. Furthermore, an unambiguous assignment of the tertiary and quaternary C-atoms of **15** and **17** was supported by analysis of the coupling patterns in the gated-decoupled ¹³C-NMR spectra (see *Exper. Part* for coupling constants, *cf.* [21]), selective heteronuclear ¹³C{¹H} decoupling and NOE experiments, and finally by a long-range heteronuclear shift-correlation experiment (HC-COSY) of **15** (*Fig. 1*).

To decide whether the exchangeable proton of compound 15 is bonded to the exocyclic N-atom (tautomer 15a) or to the ring N-atom of the dihydrothiazole ring (tautomer 15b), we irradiated at the resonance frequency of NH (9.26 ppm) in the ¹H-NMR

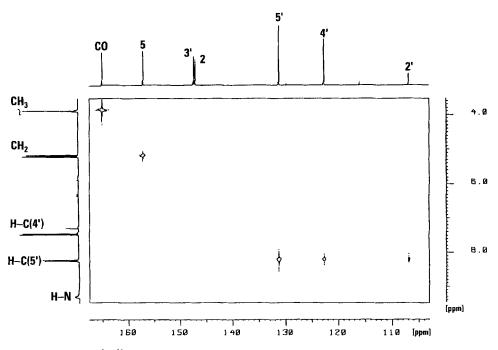


Fig. 1. 2D-1H, 13C-NMR Spectrum of 15a (250.13/62.89 MHz, long-range shift correlation)

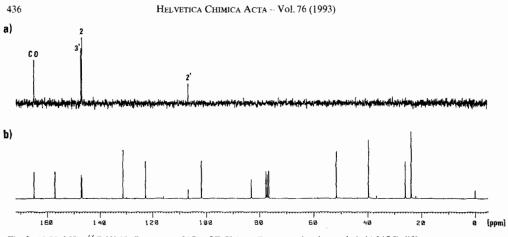


Fig. 2. 62.89-MHz 13 C-NMR Spectrum of **15a** (CDCl₃). a) Proton-noise decoupled, b) NOE-difference spectrum after irradiation of H-N.

spectrum of the compound 15, which resulted in signal enhancements of C(2'), C(3'), C(2), and $COOCH_3$ (*Fig. 2*).

An energy-optimized molecular model (ALCHEMY II) [16] of 15a (*Fig. 3*) demonstrates that the range of distances between H–N and these C-atoms is significantly smaller than in the tautomeric form 15b (15a: 1.946 Å (C(2)) to 2.644 Å ($COOCH_3$); 15b:

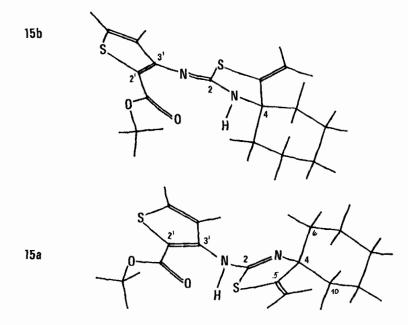


Fig. 3. Molecular model of 15a and 15b (drawing without H-atoms (= open valences) except of N-H for better overview)

1.986 Å (C(2)) to 4.359 Å (C(7))). Because of the relative short distance between H–N and C(4) in isomer **15b** (in our molecular model *ca*. 2.11 Å), the lack of a signal enhancement of C(4) may be interpreted as an argument against structure **15b**, which clearly has to be discarded.

The conclusion, that the observed NOEs do not arise from tautomer **15b** but from the tautomer **15a** is also supported by the fact that heteronuclear NOE's in ¹³C-NMR spectra are difficult to determine for distances ≥ 2.9 Å because of the critical signal/noise ratio in such cases [17]. The small signals in the ¹³C-NMR spectrum of **15a** (*Fig. 2*) are due to impurities and do not arise from tautomer **15b**. As we have found only one set of signals in the ¹³C-NMR spectra of 4, 7, and **14** too, tautomeric equilibria seem not to be existent for the presented dihydrothiazoles with the tautomer containing an exocyclic C=N bond in a detectable concentration by NMR spectroscopy.

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Experimental Part

General. M.p.: Reichert hot-stage microscope, uncorrected. IR Spectra (in KBr, cm⁻¹): Perkin-Elmer 325 and Perkin-Elmer FTIR 1600 instruments. NMR Spectra in CDCl₃: ¹H- and ¹³C-NMR: Bruker WM-250 spectrometer, 5 mm¹H/¹³C dual probehead (¹H; 250.13 MHz, ¹³C; 62.89 MHz) and Varian VXR 300 spectrometer (¹H; 300 MHz); δ in ppm rel. to TMS; multiplicities and coupling constants in ¹³C-NMR from gated-decoupled spectra (size 32 K; sweep width 7812.5 Hz (Hz/Pt = 0.477) and 7462.7 Hz (Hz/Pt = 0.455) for 17 and 15a, resp. 1 H, 13 C-COSY Spectrum: sequence: D1-90°(¹H)-D0-180°(¹³C)-D0-D3-90°(1H), 90°(¹³C)-D4-BB,FID; 90° ¹H pulse 12 µs, 180° 13 C pulse 14 µs, relaxation delay D1 = 3s, polarization time D3 = refocussing time D4 = 64.94 ms (optimized for J = 7.7 Hz), 90° spectral width 7462.7 Hz in F2, 1054.9 Hz in F1, data matrix 4096 \times 512 (zero-filling in F1), 56 scans during 256 time increments, 2 dummy scans, Lorentz to Gauss multiplication in F1 (LB = -3, GB = 0.3), exponential multiplication in F2 (LB = 1). Selective heteronuclear ${}^{13}C{}^{1}H$ -NOE Difference Spectrum (cf. [18] [19]). 80 mg/0.5 ml CDCl₃. Size: 16 K, sweep width 11111 Hz, decoupling power for selective CW pre-irradiation 51 dB below 0.2 W, decoupling powder for broad band decoupling 0 dB below 0.2 W, decoupling time 3 s, switch delay for appropriate settings of decoupler frequency, power and decoupler status twice 0.05 s, recycling time 3.837 s, pulse length 4 µs (ca. 50° pulse), 8200 acquisitions each and off-resonance (200 scans, 8 dummy scans, 41 cycles), exponential multiplication of the difference FID prior to Fourier transformation (LB = 1). EI-MS: Varian MAT 311 mass spectrometer. Elemental analyses: Heraeus C,H,N,O rapid analyzer and Perkin-Elmer 2400 CHN analyzer.

2,3-Dihydro-3,3-dimethyl-2-methylidenepyrido[3,2-d]thiazolo[3,2-a]pyrimidin-5-one (2). To a soln. of ethyl 3-isothiocyanatopyridine-2-carboxylate (1) [8] (0.28 g, 1.3 mmol) in MeOH (7 ml), 1,1-dimethylprop-2-ynylamine (0.14 g, 1.3 mmol) was added, and the mixture was heated at reflux for 6 h, precipitated product was filtered off and purified using column chromatography (silica gel, Et₂O/MeOH 3:1): 0.08 g (23%) of 3. M.p. 129–131°. IR: 2980, 2960, 1690, 1575, 1550, 1430. ¹H-NMR: 1.98 (*s*, 2CH₃); 5.31, 5.42 (2*d*, J = 2.7, CH₂=); 7.62 (*dd*, J(8,9) = 8.3, H–C(8)); 7.86 (*dd*, H–C(9)); 8.76 (*dd*, J(7,8) = 4.0, J(7,9) = 1.6, H–C(7)): ¹³C-NMR: 158.7 (C(5)); 158.1 (C(10a)); 148.7 (C(7)); 145.1 (C(9a)); 144.6 (C(2)); 137.1 (C(5a)); 133.9 (C(9)); 128.5 (C(8)); 105.6 (CH₂=); 7.33 (C(3)); 26.53 (2 CH₃). MS: 245 (100, M^+). Anal. calc. for C₁₂H₁₁N₃OS (245.31): C 58.76, H 4.52, N 17.13; found: C 59.08, H 4.25, N 16.72.

Reaction of 1 [8] and Eth-1-ynylcyclohexylamine. A soln. of 1 (0.5 g, 2.4 mmol) and eth-1-ynylcyclohexylamine (0.311 g, 2.53 mmol) in MeOH (7 ml) was heated under reflux for 8 h, evaporated under reduced pressure and separated by column chromatography (silica gel, $CH_2Cl_2/acetone$, 100:1 and 30:1, as solvents). The first

fraction gave, after evaporation, *ethyl* 3-[(4-methylidene-3-thia-1-azaspiro[4.5]dec-1-en-2-yl)amino]pyridine-2carboxylate (**3**; 0.064 g, 8.1%). M.p. 106–110°. IR: 3241, 2927, 2854, 1684, 1639, 1587, 1516, 1456, 1389, 1313, 1255, 1206, 1154, 1091, 1033, 992, 856. ¹H-NMR: 1.2–1.87 (*m*, cyclohexylidene); 1.4 (*t*, J = 7.2, CH₃); 4.43 (*q*, J = 7.2, CH₂O); 5.09, 5.12 (2*d*, J = 1.9, CH₂=); 7.41 (*dd*, J(5,6) = 4.6, H–C(5)); 8.26 (*dd*, H–C(6)); 9.22 (*dd*, J(4,5) = 8.7, J(4,6) = 1.2, H–C(4)); 10.24 (br., NH). ¹³C-NMR: 167.8 (COOEt); 156.1 (C(6)); 147.7 (C(2')); 141.5 (C(4)); 116.1 (C(5)); 101.9 (CH₂=); 62.2 (CH₂O); 39.6 (C(6,10)); 25.9 (C(8)); 23.7 (C(7,9)); 14.2 (CH₃). MS: 331 (100, M^+). Anal. calc. for C₁₇H₂₁N₃O₂S (331.42): C 61.60, H 6.39, N 12.68; found: C 61.28, H 6.19, N 12.60.

The second fraction gave, after evaporation, 2',3'-dihydro-2'-methylidenespiro[cyclohexane-1,3'-pyrido-[3,2-a]thiazolo[3,2-d]pyrimidin]-5'-one (4; 0.134 g, 25.5%). M.p. 144–146° (from EtOH). IR: 3360, 2998, 1695, 1573, 1550, 1462, 1423, 1361, 1319, 1298, 1220, 1205, 1127, 1010, 885, 820, 810, 695. ¹H-NMR: 1.5–2.02 (*m*, 8H, cyclohexylidene); 3.08-3.29 (*m*, 2H, cyclohexylidene); 5.32, 5.78 (2d, J = 2.6, CH₂=); 7.59 (dd, J(8',9') = 8.3, H–C(8')); 7.83 (dd, H–C(9')); 8.74 (dd, J(7',8') = 4.3, J(7',9') = 1.6, H–C(7')). ¹³C-NMR: 159.0 (d, J = 1.3, C(5')); 158.7 (*s*, C(10'a)); 148.6 (ddd, J = 182.3, 7.9, 3.3, C(7')); 144.5 (dt, J = 8.4, 1.7, C(9'a)); 144.0 (*m*, C(2')); 137.6 (ddd, J = 11.7, 4.6, 1.3, C(5'a)); 133.8 (dd, J = 166.8, 6.3, C(9')); 128.5 (dd, J = 165.6, 9.6, C(8')); 108.8 (dd, J = 163.5, 161.4, CH₂=); 76.1 (br. *s*, C(3')); 30.3 (C(1), C(6)); 23.3 (C(4)); 21.7 (C(3), C(5)). MS: calc. for C₁₅H₁₅N₃OS: 285.0936; found: 285.0935.

Under the same reaction conditions in benzene only compound 3 was formed in 62.8% yield.

2,3-Dihydro-3,3-dimethyl-2-methylidene-5-pyrido[2,3-d]thiazolo[3,2-a]pyrimidin-5-one (6). Prepared in the same manner as compound **2** in MeOH: 44.2 % (from EtOH). M.p. 143–145°. IR: 3420, 3000, 1689, 1575, 1509, 1430, 1329, 1286, 1240, 1179, 1140, 1011, 895, 794. ¹H-NMR: 1.95 (s, 2 CH₃); 5.31, 5.41 (2d, J = 1.8, CH₂=); 7.35 (dd, J(7,8) = 4.8, H–C(7)); 8.53 (dd, J(6,7) = 7.8, J(6,8) = 2.1, H–C(6)); 8.88 (dd, H–C(8)). ¹³C-NMR: 161.3 C(5)); 160.4 (C(10a)); 157.9 (C(9a)); 156.1 (C(8)); 144.0 (C(2)); 136.3 (C(6)); 121.3 (C(7)); 115.3 (C(5a)); 105.6 (CH₂=); 73.2 (C(3)); 26.7 (2 CH₃). MS: 245 (100, M^+). Anal. calc. for C₁₂H₁₁N₃OS (245.31): C 58.76, H 4.52, H 17.13; found: C 58.78, H 4.62, N 17.09.

Reaction of Ethyl 2-Isothiocyanatopyridine-3-carboxylate (5) [11] *and Eth-1-ynylcyclohexylamine*. A soln. of 5 (0.3 g, 1.44 mmol) and eth-1-ynylcyclohexylamine (0.201 g, 1.63 mmol) in benzene or EtOH (15 ml) was heated under reflux for 4 h, evaporated under reduced pressure and separated by column chromatography (silica gel, CHCl₃/Acetone 50:1 and 30:1, as solvents). The first fraction gave, after evaporation, *ethyl 2-[(4-methylidene-3-thia-1-azaspiro[4.5]dec-1-en-2-yl)amino]pyridine-3-carboxylate* (7, 0.253 g, 52.8%). M.p. 145–146° (from EtOH). IR: 3210, 2940, 2930, 2860, 1698, 1628, 1605, 1582, 1510, 1460, 1450, 1395, 1370, 1310, 1290, 1260, 1145, 1090, 1072, 1025, 990, 842, 779. ¹H-NMR: 1.2–1.92 (*m*, cyclohexylidene); 1.38 (*t*, *J* = 7.1, CH₃); 4.37 (*q*, *J* = 7.1, CH₂); 5.02, 5.16 (2*d*, *J* = 1.3, CH₂=); 6.84 (d*d*, *J*(5,6) = 4.8, H–C(5)); 8.21 (d*d*, *J*(4,5) = 7.8, *J*(4,6) = 1.6, H–C(4)); 8.34 (d*d*, *J*(5,6) = 4.8, H–C(6)); 10.7 (br. *s*, NH); ¹³C-NMR: 166.6 (*m*, COOEt); 157.1 (d*d*, *J* = 179.5, 7.6, 3.6, C(6)); 148.7 (overlap in gated dec. spectrum, C(2')); 140.2 (d*d*, *J* = 165.2, 6.7, 1.8, C(4)); 115.6 (d*d*, *J* = 167.9, 7.6, C(5)); 108.1 (*m*, C(3)); 100.8 (*t*, *J* = 160.5, CH₂=); 75.8 (br., C(4')); 61.7 (CH₂O); 39.7 (C(6'), C(10')); 26.0 (C(8')); 23.1 (C(7'), C(9')); 14.2 (CH₃). MS: 331 (100, *M*⁺). Anal. calc. for C₁₇H₂₁N₃O₂S (331.42); C 61.60, H 6.39, N 12.68; found: C 61.77, H 6.53, N 12.77.

The second fraction gave, after evaporation, *ethyl* 2-[(3-thia-1-azaspiro[5.5]undeca-1,4-dien-2-yl)amino]pyridine-3-carboxylate (**8**; 29.8%). M.p. 86–87°. IR: 3302, 2930, 2855, 1693, 1618, 1589, 1499, 1451, 1421, 1368, 1328, 1288, 1256, 1182, 1136, 1086, 991, 923, 906, 814. ¹H-NMR: 1.4 (t, J = 7.1, CH₃); 1.3–1.9 (m, cyclohexylidene); 4.38 (q, J = 7.1, CH₂O); 5.85, 6.19 (2d, J = 9.6, H–C(4'), H–C(5')); 6.84 (dd, J(5,6) = 4.8, H–C(5)); 8.19 (dd, J(4,5) = 7.6, J(4,6) = 1.7, H–C(4)); 8.34 (dd, H–C(6)); 10.28 (br., NH). MS: 331 (73, M^+), 28,8 (100, [M – 43]⁺). Anal. calc. for C₁₇H₂₁N₃O₂S (331.42): C 61.60, H 6.39, N 12.68; found: C 61.47, H 6.23, N 12.46.

Reaction of Methyl 3-Isothiocyanatopyrazine-2-carboxylate (9)[10] and 1,1-Dimethylprop-2-ynylamine. A soln. of 9 (0.42 g, 2.2 mmol) and 1,1-dimethylprop-2-ynylamine (0.26 g, 2.4 mmol) in MeOH (5 ml) was heated under reflux for 6 h. Precipitated product was filtered off and recrystallized from EtOH to give 0.33 g (62%) of 2,3-dihydro-8,8-dimethyl-7-methylidenethiazolo[2,3-b]pteridin-10-one (10a) and 9,9-dimethylthiazino[2,3-b]pteridin-11-one (11a) in a ratio (according to NMR data) of 1:1.2. It was not possible to separate both products. M.p. 233–235°. IR: 3440, 2990, 1708, 1575, 1519, 1450, 1410, 1320, 1283, 1235, 1210, 1160, 1130, 1110, 910, 880, 825, 725. MS: 246 (100, M^{++}). Anal. calc. for C₁₁H₁₀N₄OS (246.29): C 53.64, H 4.09, N 22.75; found: C 53.52, H 4.04, N 22.60.

10a: ¹H-NMR: 1.99 (*s*, 2 CH₃); 5.34, 5.45 (2*d*, J = 2.9, CH₂=); 8.66, 8.86 (2*d*, J = 2.1, H–C(2), H–C(3)). **11a**: ¹H-NMR: 1.94 (*s*, 2 CH₃); 5.70, 6.09 (2*d*, J = 10.7, H–C(8), H–C(7)); 8.69, 8.86 (2*d*, J = 2.1, H–C(2), H–C(3)).

Reaction of 9 [10] and Eth-1-ynylcyclohexylamine. A soln. of 9 (0.45 g, 2.3 mmol) and eth-1-ynylcyclohexylamine (0.28 g, 2.3 mmol) in EtOH (5 ml) was heated under reflux for 6 h. Precipitated product was filtered off and recrystallized from i-PrOH to give 0.13 g (32%) of 2',3'-dihydro-7'-methylidenespiro[cyclohexane-1,8'-thiazolo[2,3-b]pteridin]-10'-one (10b) and spiro[cyclohexane-1,9'-thiazino[2,3-b]pteridin]-11'-one (11b) in a ratio of 3.6:1 (NMR). It was not possible to separate the products. M.p. 217–220°. IR: 3440, 2939, 2870, 1700, 1582, 1550, 1540, 1460, 1405, 1325, 1269, 1249, 1210, 1190, 1115, 1040, 900, 830, 790, 725. ¹H-NMR: 1.47–2.08 (*m*, 8 H, cyclohexylidene); 3.0–3.22 (*m*, 2 H, cyclohexylidene); 5.36, 5.80 and 6.24, 6.42 (2 × 2d, first set of signals J = 2.8, CH₂=, second set of signals J = 10.6, H–C(7'), H–C(8'), **11b**; 8.69, 8.84 (2d, J = 1.2, pyrazine). MS: 286 (100, M^{++}). Anal. calc. for C₁₄H₁₄N₄OS (286.36): C 58.72, H 4.93, N 19.57; found: C 58.95, H 4.99, N 19.68.

2,3-Dihydro-7,7-dimethyl-6-methylidenethiazolo[3,2-a]thieno[3,2-d]pyrimidine (13). From methyl 3-isothiocyanatothiophene-2-carboxylate (12; 0.32 g, 1.61 mmol) and 1,1-dimethylprop-2-ynylamino (0.15 g, 1.61 mmol) in MeOH (8 ml) at reflux for 6 h. Precipitated product was filtered off and crystallized from EtOH: 0.24 g (60.6%) of 13. M.p. 134–135°. IR: 3450, 3130, 2980, 1680, 1550, 1500, 1305. ¹H-NMR: 1.97 (s, 2 CH₃); 5.28, 5.39, (2d, J = 1.7, CH₂=); 7.16 (d, J = 5.1, H–C(3)); 7.71 (d, J = 5.1, H–C(2)). MS: 250 (100, M^+). Anal. calc. for C₁₁H₁₀N₂OS₂ (250.34): C 52.78, H 4.03, N 11.19; found: C 52.65, H 4.03, N 11.12.

Methyl 3-[(4,5-Dihydro-4,4-dimethyl-5-methylidenethiazol-2-yl)amino]thiophene-2-carboxylate (14). A soln. of 12 (0.46 g, 2.3 mmol) and 1,1-dimethylprop-2-ynylamine (0.23 g, 2.8 mmol) in benzene (10 ml) was heated under reflux for 6 h, evaporated under reduced pressure, and purified by column chromatography (silica gel, CHCl₃/acetone 50:1): 0.42 g (64.5%) of 14. M.p. 99–100°. IR: 3310, 2967, 2854, 1675, 1632, 1572, 1452, 1425, 1260, 1165, 1105, 1023, 971, 875, 780. ¹H-NMR: 1.46 (*s*, 2CH₃); 3.87 (*s*, CH₂O); 5.14, 5.16 (2*d*, J = 1.8, CH₂=); 7.41 (*d*, J = 5.5, H–C(4)); 8.11 (*d*, J = 5.5, H–C(5)); 9.2 (br., NH). ¹³C-NMR: 165.0 (COOMe); 156.6 (C(5')); 148.3 (C(3)); 147.2 (C(2')); 131.3 (C(5)); 122.6 (C(4)); 106.9 (C(2)); 102.4 (CH₂=); 80.0 (C(4')); 51.7 (CH₃); 30.2 (2 CH₃). MS: calc. for C₁₂H₁₄N₂O₂S₂: 282.0497; found: 282.0496.

Methyl 3-[(4-Methylidene-3-thia-1-azaspiro[4.5]dec-1-en-2-yl)amino]thiophene-2-carboxylate (**15a**). A soln. of **12** (0.51 g, 2.6 mmoles) and eth-1-ynylcyclohexylamine (0.32 g, 2.6 mmol) in EtOH or benzene (10 ml) was heated under reflux for 6 h. Precipitated product was filtered off and recrystallized from EtOH: 0.63 g (84.8%) of **15a**. M.p. 139–142°. IR: 3320, 2949, 2859, 1685, 1670, 1640, 1612, 1572, 1451, 1420, 1408, 1281, 1255, 1160, 1105, 999, 872, 780, 730. ¹H-NMR: 1.30–1.98 (*m*, cyclohexylidene); 3.88 (*s*, CH₃); 5.12, 5.19 (2*d*, J = 1.8, CH₂=); 7.42 (*d*, J = 5.6, H–C(4)); 8.20 (*d*, J = 5.6, H–C(5)); 9.28 (br., NH). ¹³C-NMR: 165.0 (*m*, COOMe); 157.1 (br. *s*, C(4')); 147.3 (*dd*, J = 11.4, 2.3, C(3)); 147.0 (*s*, C(2')); 131.3 (*dd*, J = 185.8, 7.3, C(5)); 122.6 (*dd*, J = 177.6, 4.6, C(4)); 106.8 (*dd*, J = 7.7, 5.5, C(2)); 101.9 (*dd*, J = 162.2, 159.9, CH₂=); 83.1 (br. *s*, C(5')); 51.7 (CH₃); 39.7 (C(6'), C(10')); 26.0 (C(8')); 23.8 (C(7'), C(9')). MS: 322 (100, M^+). Anal. calc. for C₁₅H₁₄N₂O₂S₂ (322.45): C 55.87, H 5.63, N 8.69; found: C 55.78, H 5.61, N 8.63.

6',7'-Dihydro-6-methylidenespiro[cyclohexane-1,7'-thiazolo[3,2-a]thieno[3,2-d]pyrimidin]-9'-one (16). To a soln. of 15 (0.64 g, 2 mmol) in abs. toluene (10 ml), t-BuOK (0.09 g, 0.7 mmol) was added, the mixture stirred at reflux temp. for 6 h, and then washed with 10 ml of H₂O. The org. phase was dried (Na₂SO₄). After drying, the solvent was removed under reduced pressure and the product recrystallized from EtOH: 0.35 g (60.1%) of 16. M.p. 98–100°. IR: 3085, 2929, 1658, 1544, 1500, 1452, 1270, 1222, 1125, 1015, 876, 782. ¹H-NMR: 1.42–2.0 (*m*, 8H, cyclohexylidene); 3.05–3.25 (*m*, 2H, cyclohexylidene); 5.29, 5.74 (2*d*, J = 1.8, CH₂=); 7.15 (*d*, J = 5.3, H–C(3')); 7.71 (*d*, J = 5.3, H–C(2')). MS: 290 (100, M^+). Anal. calc. for C₁₄H₁₄N₂OS₂ (290.41): C 57.90, H 4.86, N 9.65; found: C 58.16, H 4.95, N 9.69.

Methyl 3-Aminothiophene-2-carboxylate (17). Commercial product available from *Aldrich* was used without further purification. ¹H-NMR: 3.81 (*s*, CH₃); 5.53 (br. *s*, NH₂); 6.51 (*d*, J = 5.4, H–C(4)); 7.24 (*d*, J = 5.4, H–C(5)). ¹³C-NMR: 164.8 (*m*, COOMe); 154.0 (*dd*, J = 10.5, 3.8, C(3)); 131.4 (*dd*, J = 184.5, 5.7, C(5)); 119.7 (*dt*, J = 5.3, 4.3, C(4)); 100.9 (*m*, C(2)); 51.5 (*q*, J = 146.9, CH₃).

REFERENCES

- J. V. Metzger, in 'Comprehensive Heterocyclic Chemistry', Eds. A. R. Katritzky and C. W. Rees, Pergamon Press, Oxford, 1984, Vol. 6, pp. 235–331.
- [2] F. Sauter, W. Deinhammer, Monatsh. Chem. 1974, 105, 270, 452.
- [3] F. Kienzle, to F. Hoffmann-La Roche AG, EP 0043054, 1982.
- [4] F. Hiroshi, S. Toshiaki, S. Mitsuo, T. Keiichi, JP 63,225,383, 1988.
- [5] Y. Yura, Chem. Pharm. Bull. 1962, 10, 372.
- [6] N.R. Easton, D.R. Cassady, R.D. Dillard, J. Org. Chem. 1964, 29, 1851.
- [7] P. Thieme, H. Koenig, Synthesis 1973, 426.

- [8] U. Urleb, B. Stanovnik, M. Tišler, J. Heterocycl. Chem. 1990, 27, 407.
- [9] U. Urleb, B. Stanovnik, M. Tišler, J. Heterocycl. Chem. 1990, 27, 413.
- [10] U. Urleb, R. Neidlein, W. Kramer, J. Heterocycl. Chem. 1990, 27, 433.
- [11] U. Urleb, B. Stanovnik, M. Tišler, J. Heterocycl. Chem. 1990, 27, 643.
- [12] J.E. Baldwin, J. Chem. Soc., Chem. Commun. 1976, 734.
- [13] J.I. Dickstein, S.I. Miller, in 'The Chemistry of the carbon-carbon triple bond, Ed. S. Patai, John Wiley and Sons, New York, 1978, Part 2, p.827.
- [14] A. R. Katritzky, J.M. Lagowski, in 'Advances in Heterocyclic Chemistry', Ed. A. R. Katritzky, Academic Press, New York, 1963, Vol. 1, p. 339.
- [15] M. Brnix, J. De Mendoza, R. M. Claramont, J. Elguero, Magn. Reson. Chem. 1985, 23, 367.
- [16] Tripos Associates, Inc., St. Louis, Missouri, USA.
- [17] H. R. Loosli, H. Kessler, H. Oschkinat, H. P. Weber, T. J. Petcher, A. Widmer, Helv. Chim. Acta 1985, 68, 682.
- [18] R. Neidlein, W. Kramer, V. Ullrich, Helv. Chim. Acta 1986, 69, 898.
- [19] C. Cativiela, F. Sanchez-Ferrando, Magn. Reson. Chem. 1985, 23, 1072.